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A type of bond that holds cations and

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anions together A type of bond in which two atoms share a pair of valence electrons

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Chapter 6- Chemical Bonds. 6.1 Ionic Bonding 6.2 Covalent Bonding 6.3 Naming Compounds and Writing Formulas 6.4 The Structure of Metals. STUDY. PLAY. Electron Dot Diagram. a model of an atom in which each dot represents a valence electron. Ion. a particle that is electrically charged (positive or negative)

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The Structure of Metals (pages 176-181)  
This section discusses metallic bonds  
and the properties of metals. It also  
explains how the properties of an alloy  
are controlled. <http://mrmicco.weebly.com/uploads/1/2/5/8/12584959/psgs1064b.pdf>

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based on a set of properties that repeat  
from ... Chapter 5 The Periodic Table  
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cations and anions together. Covalent. A  
type of bond in which two atoms share a  
pair of valence electrons. Molecule.  
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