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Solution From the first law of thermodynamics applied for a closed system, the heat transferred is given by:

$Q = \Delta U$ The change in internal energy is defined by Equation

8.3-6 of Felder & Rousseau: $\Delta U = \int_{T_1}^{T_2} C_v(T) dT$

Substituting the heat capacity equation into the integral: $\int_{T_1}^{T_2} C_v(T) dT$

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